

## Solution to the Reflective Homework

- 1) Since the potential energy is infinite for  $x < 0$ , the possible wave function for the system must be zero at  $x = 0$ . The solutions for the time-independent Schroedinger equation for a half harmonic oscillator are a subset of the solutions for the full harmonic oscillator (since the problem for  $x > 0$  is identical for both full and half harmonic oscillators). This means that the energies for a half harmonic oscillator must be the odd  $n$  energies of the corresponding full harmonic oscillator (recall that the even quantum numbers denoting energies of the full harmonic oscillator  $n=0,2,4,6...$ etc. do not have stationary state wave functions that go to zero at  $x = 0$ ).
- 2) For both the infinite square well and the simple harmonic oscillator, in the limit as  $n$  goes to infinity, the difference in the energy per unit energy is zero. For example,  $\lim_{n \rightarrow \infty} (E_{n+1} - E_n)/E_n = \lim_{n \rightarrow \infty} (2n+1)/n^2 = \lim_{n \rightarrow \infty} 2/n = 0$  for the infinite square well and  $\lim_{n \rightarrow \infty} (E_{n+1} - E_n)/E_n = \lim_{n \rightarrow \infty} 1/n = 0$  for a harmonic oscillator. This result is one manifestation of the correspondence principle. The discreteness of the energy levels is not important for large  $n$  states and the results of quantum mechanics will tend towards the classical results.
- 3) The method for finding the wave function at time  $t$  for a non-stationary state wave function for a harmonic oscillator and for an infinite square well are similar. You should first write the wave function for the system at time  $t = 0$  in terms of a linear superposition of stationary states. Then, at time  $t$ , each of the terms in the superposition will develop a time-dependent phase factor  $e^{-iE_n t/\hbar}$ .
- 4) The expectation values of none of these observables (or any time-independent observable) depend on time in a stationary state. That is a defining property of a stationary state.